LITTLE FLOWER COLLEGE DEPARTMENT OF CHEMISTRY

TOPIC : HERMITIAN OPERATOR AND POSTULATES

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HERMITIAN OPERATORS

- Using Quantum mechanical principles we can predict theoretically the value of physical observable .
- When we measure that property we are actually getting the eigen value of the operator associated with the property.
- The eigen values can be real or imaginary
- But we cannot make an imaginary measurement.
- So we have to develop an operator which will yield only real values . Such operator are known as Hermitian operator.
- The eigen function of a hermitian operator are orthogonal.
- All quantum mechanical operators are hermitian operator such as x, Px, Vx , Tx, Hx.

•Let A^ be the linear operator representing the physical property of A and it has two eigen function

$$Ψ*$$
 and $ψ$. The average value of A is
= \$\int ψ *A ψd τ\$

Since the average value of a physical quantity must be a real number, we demand that

The linear operator that satisfies for all well behaved functions is called a Hermitian operator.

PROPERTIES OF HERMITIAN OPERATOR

- THEOREM 1:
- Eigen values of hermitian operator are real numbers.

For Hermitian operator Â

$$\int (\hat{A}\Psi)^* \Psi \, d\tau = \Psi^* (\hat{A}\Psi) d\tau \rightarrow \mathbf{1}$$
From Eigen value equation

$$\hat{A}\Psi = a\Psi \rightarrow \mathbf{2}$$
Using **2** & **1**

$$\int (\alpha\Psi)^* \Psi \, d\tau = \int \Psi^* (\alpha\Psi) \, d\tau$$

$$a^* \int \Psi^* \Psi \, d\tau = a \int \Psi^* \Psi \, d\tau$$

$$\int \Psi^* \Psi \, d\tau = constant$$

$$\int \Psi^* \Psi \, \tau \neq \mathbf{0}$$
Hence $(a^* - a) \int \Psi^* \Psi \, d\tau = \mathbf{0}$

$$a^* - a = \mathbf{0}$$

$$a^* = a \quad a \text{ is real}.$$

THEOREM 2:

- The eigen function corresponding to two different eigen values for hermitian operator are orthogonal
- Let A ^ is hermitian operator and we have two different eigen function ψm, ψn corresponding to two different eigen values am, an , then from eigen value equation

$$\hat{A}\Psi_{m} = a_{m}\Psi_{m} \rightarrow \mathbf{1}$$

 $\hat{A}\Psi_n = a_n\Psi_n \rightarrow 2$

From the definition of hermitian operator

$$\int (\hat{A}\Psi_n)^* \Psi_m \, d\tau = \int \Psi_n^* (\hat{A}\Psi_m) \, d\tau$$

Using
$$\blacksquare$$
 & $\textcircled{2}$ In this equation

$$\int (a_n \Psi_n)^* \Psi_m d\tau = \int \Psi_n^* (a_m \Psi_m) d\tau$$

$$a_n^* \int \Psi_n^* \Psi_m d\tau = a_m \int \Psi_n^* \Psi_m d\tau$$

$$a_n^* = a_n$$

$$a_n - a_m \int \Psi_n^* \Psi_m d\tau = 0$$

$$a_n \neq a_m \Rightarrow a_n - a_m \neq 0$$

$$\int \Psi_n^* \Psi_m d\tau = 0$$

EIGEN VALUE POSTULATE

- This tells us about the result of the measurement of variable .
- The result of measurement of an observable A will always be one of the eigen value of the operator associated with the property as given by the eigen value equation. $\hat{A}\phi_n = a_n \phi_n$ and it need not be the state function .
- If the eigen function becomes the state function we can make precise measurement.

This postulate connect the theory and experiment.

- The eigen function and the eigen value of a system can be obtained by solving the equation .
- But here an=a1,a2,a3..... So that we get a spectrum of value ie, the theory doesn't tell about the accuracy.
- If we want to obtain concordant value we have to apply some condition .
- Here the eigen values must be real since they represent observable physical quantity.
- So the operators must be a hermitian one which yield only real eigen values

•As a example , consider the measurement of energy. The operator corresponding to the energy is the hamiltonian

Operator and its eigen equation is

$$\widehat{H}\psi = E\psi$$

- This is just a schrondinger equation the solution of this equation gives the ψn and En.
- For the case of a particle in a box En = n2h2/8ma2
- Postulate 3 says that if we measure the energy of a particle in a box we shall find one of these energies and no other.

EXPECTATION / MEAN VALUE POSTULATES

- According to third postulate when we measure a definite value we get a spectrum of value then we have to go for an average.
- The fourth postulate states that in a series of measurement of an observable A of a system described by the state function ψn a distribution of eigen value will be obtained.
- The average / mean / expectation value of which is given by the expression.

Value of which is given by the expression $\langle \bar{A} \rangle = \bar{a}_n$

$$= \frac{\int \Psi_{\mathbf{n}}^* \hat{A} \Psi_{\mathbf{n}} d\tau}{\int \Psi_{\mathbf{n}}^* \Psi_{\mathbf{n}} d\tau} \rightarrow \frac{\int \Psi_{\mathbf{n}}^* a_{\mathbf{n}} \Psi_{\mathbf{n}} d\tau}{\int \Psi_{\mathbf{n}}^* \Psi_{\mathbf{n}} d\tau}$$

If
$$\Psi_n$$
 is eigen function of \hat{A} , then $\bar{a}_n = a_n \rightarrow \mathbf{1}$

Ie. the expection value I avg value is an eigen value itself

Taking square of the observable

$$\begin{aligned} <\hat{A}^{2}> &= \bar{a}_{n}^{2} = \frac{\int_{all space} \Psi_{n}^{*} \hat{A}^{2} \Psi_{n} d\tau}{\int_{all space} \Psi^{*} \Psi d\tau} \\ &= \frac{\int_{all space} \Psi_{n}^{*} \hat{A} (\hat{A} \Psi_{n}) d\tau}{\int_{all space} \Psi^{*} \Psi d\tau} \\ &= \frac{\int \Psi_{n}^{*} \hat{A} (a_{n} \Psi_{n}) d\tau}{\int \Psi^{*} \Psi d\tau} \end{aligned}$$

$$= \frac{a_n \int \Psi_n^* (\hat{A} \Psi_n) d\tau}{\int \Psi^* \Psi d\tau}$$

$$= \frac{a_n \int \Psi_n^* a_n \Psi_n d\tau}{\int \Psi^* \Psi d\tau}$$

$$= \frac{a_n^2 \int \Psi_n^* \Psi_n d\tau}{\int \Psi^* \Psi d\tau}$$
So, $\bar{a}_n^2 = a_n^2 \rightarrow 2$
From **1** & **2**
 $\bar{a}_n^2 = a_n^2 \rightarrow (\bar{a}_n)^2$
 $[(\bar{a}_n)^2 \rightarrow \text{avg. of the square}]$

Average of square = square of the average

- This is true only when all the measurement give rise to the same result ie, a1= a2= a3.....=an
- This condition is true when the state function ψn is an eigen function of the operator A. It is very difficult to generate a state function which is an eigen function too.
- For every system there is a state function ψ corresponding to the operator which is also an eigen function of H.
- So we have $H \psi = E \psi$, therefore precise measurement of energy of a system is possible.

THANK YOU