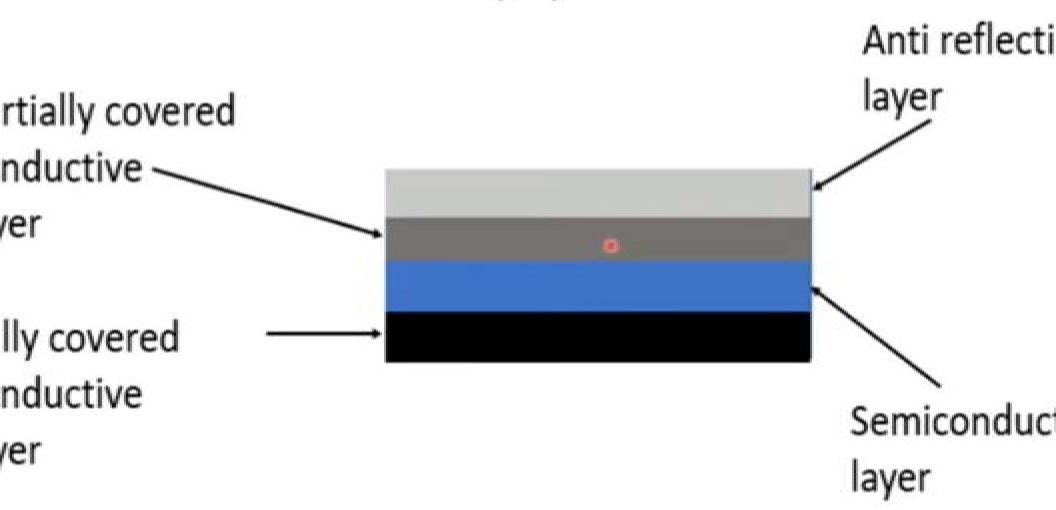
PHOTOVOLTAIC CELL

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photo-voltaic cell is an energy harvesting device, that converts sola nergy into electricity through a process called the photovoltaic effe he photovoltaic effect is a process of generating electrical voltage of lectric current from a photovoltaic cell when it is exposed to sunlig

nstruction

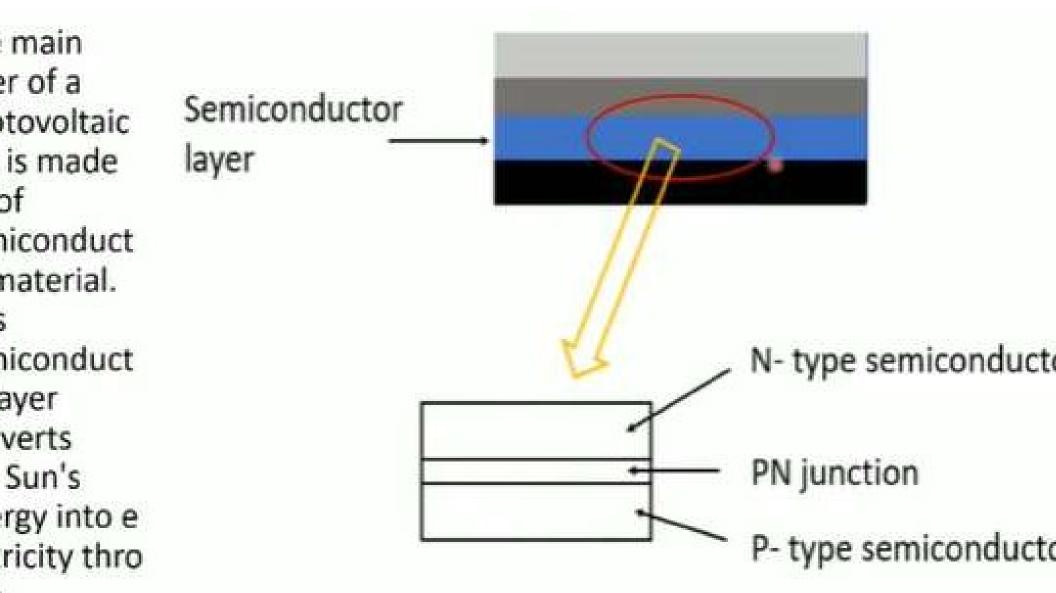
hotovoltaic cell consists of many layers of materials.



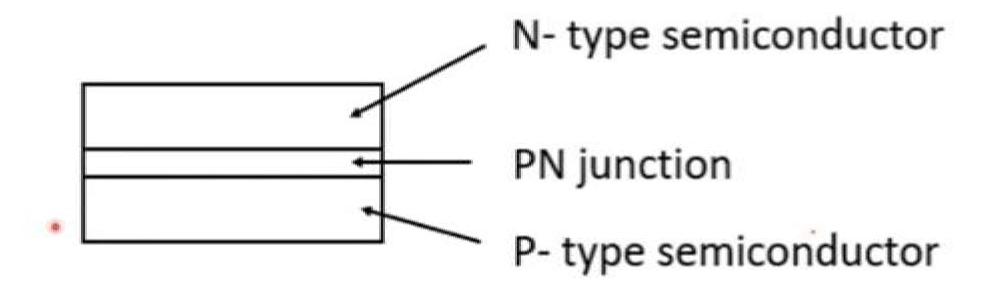
main er of a tovoltaic is made niconduct naterial.

niconduct ayer verts Sun's rgy into e

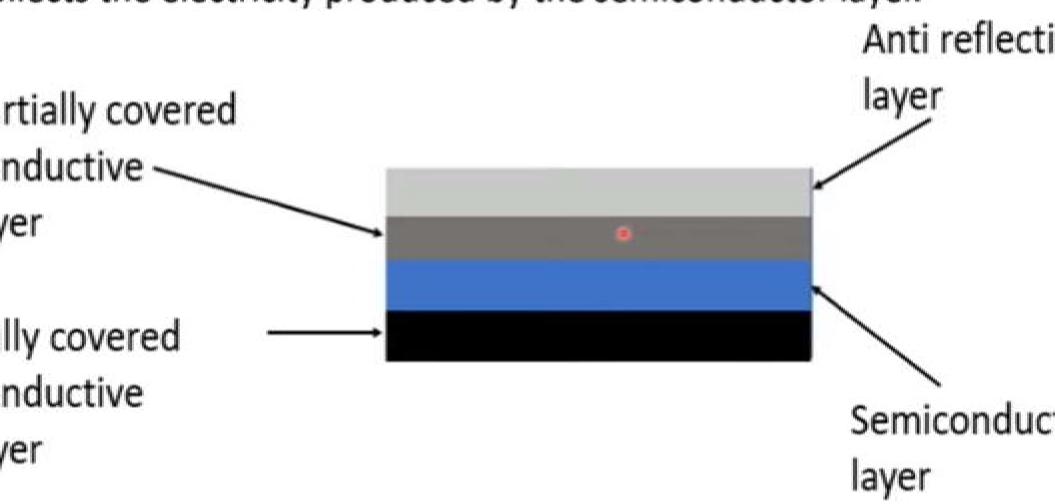
tovoltaic ct.



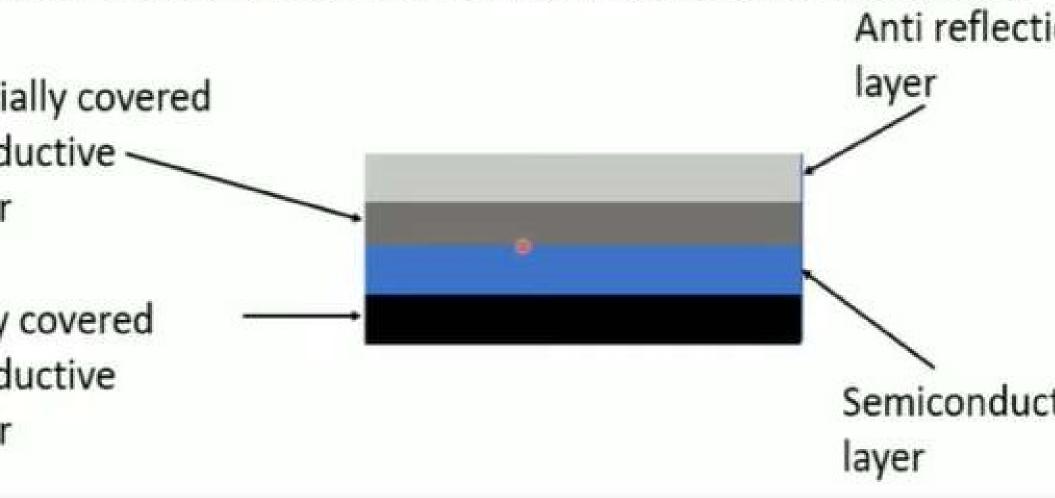
he semiconductor layer is comprised of two distinct layers of p-type and n-ty emiconductor materials.

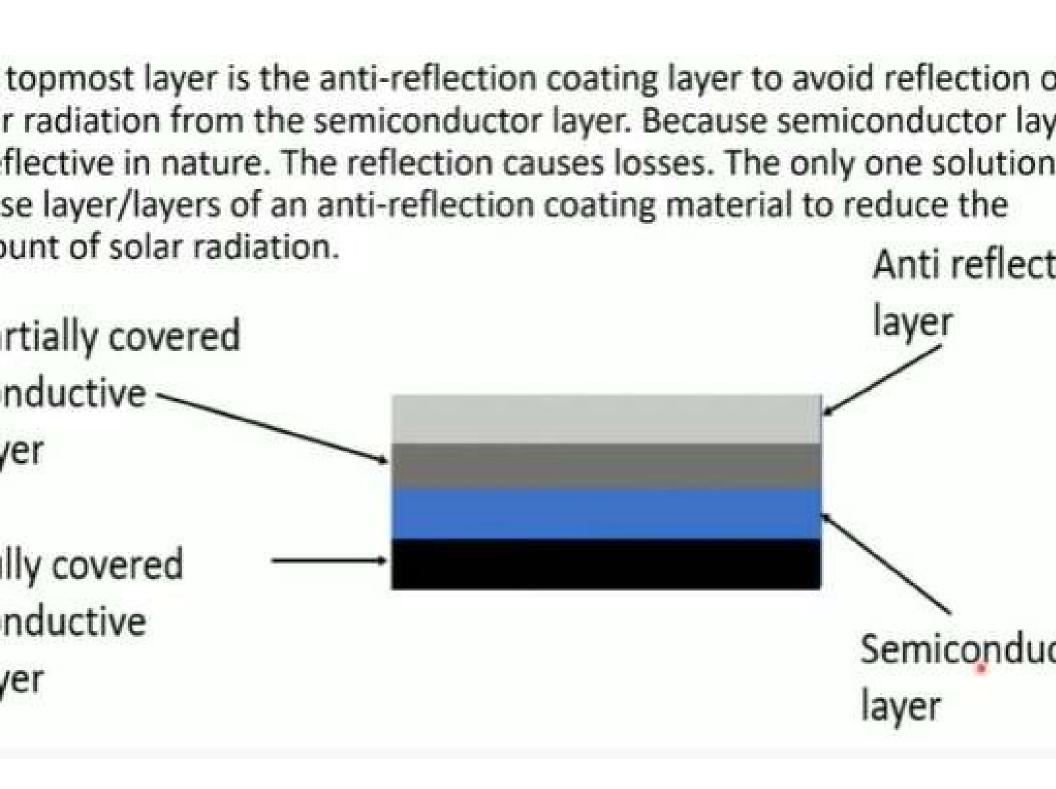


layer of conducting material is placed on the either side of the miconductor layer. The function of the conducting layer is to electricity produced by the semiconductor layer.



noted that the backside of the semiconductor layer cell completely ered with the conducting material, whereas the illuminated side of the iconductor partially covered with conducting material to avoid blocki much of the Solar radiation from reaching the semiconductor layer.





/orking

The semiconductor layer solar cell is composed of two different typ of semiconductor layers.

A p-type layer and an n-type layer that are joined together to create properties provided the properties of the properti

ght is composed of small bundles of electromagnetic radiation eneralled photons. When light energy of a suitable wavelength is incider a photovoltaic cells, the energy from the photon is transferred to an ectron of the semiconducting material. So electrons of atoms get afficient energy to jump to a higher energy state. This higher energy ate is known as the conduction band and the electrons are known see electrons. These electrons are free to move through the material of it is this motion of the electron that creates an electric current in esolar cell.

solar photovoltaic cell generates electricity from sun light through following ree general steps:

ight energy hit on the semiconductor layer and knocks electrons from atoms ree electrons.

creating a current flow inside the solar cell.

his current is captured and transferred to out side the solar cell.

Solar Cell Efficiency

There are many factors that limit the efficiency of a solar cell. The semiconductors requires a minimum photon energy to excite an electron from its parent atom as a free electron. This energy is known as the band-gap energy. If a photon has less energy than the band-gap energy, the photon is absorbed as thermal energy by the atom. In the case of silicon, the band-gap energy is 1.12 electron volts.

- ne photon energy from the sun cover a wide range of energies.
- ome of the incoming energy from the Sun does not have sufficient nergy to knock off an electron from the semiconductor atom.
- he energy below the band gap energy is absorbed by the atoms and nergy will be transferred to heat.
- ven some of energies above the band-gap energy will be also ansformed into heat.
- his reduces the efficiency again because that heat energy is not being sed to produce electricity.
- the theoretical efficiency of silicon solar cells is about 33%.

There are ways to improve the efficiency of solar cells.

ncreasing the purity of the semiconductor

Jse Gallium Arsenide instead of silicon since Gallium Arsenide is nore efficient than silicon.

lowever all of these methods are costly.

Solar cell construction materials

The most common material for commercial solar cell construction is Silicon, Gallium Arsenide, Cadmium Telluride and Copper Indium Gallium Selenide.