

Biochemistry

Metabolism of Carbohydrates

Dr Jilna Alex N

METABOLISM

GLYCOGENESIS – Biosynthesis of glycogen

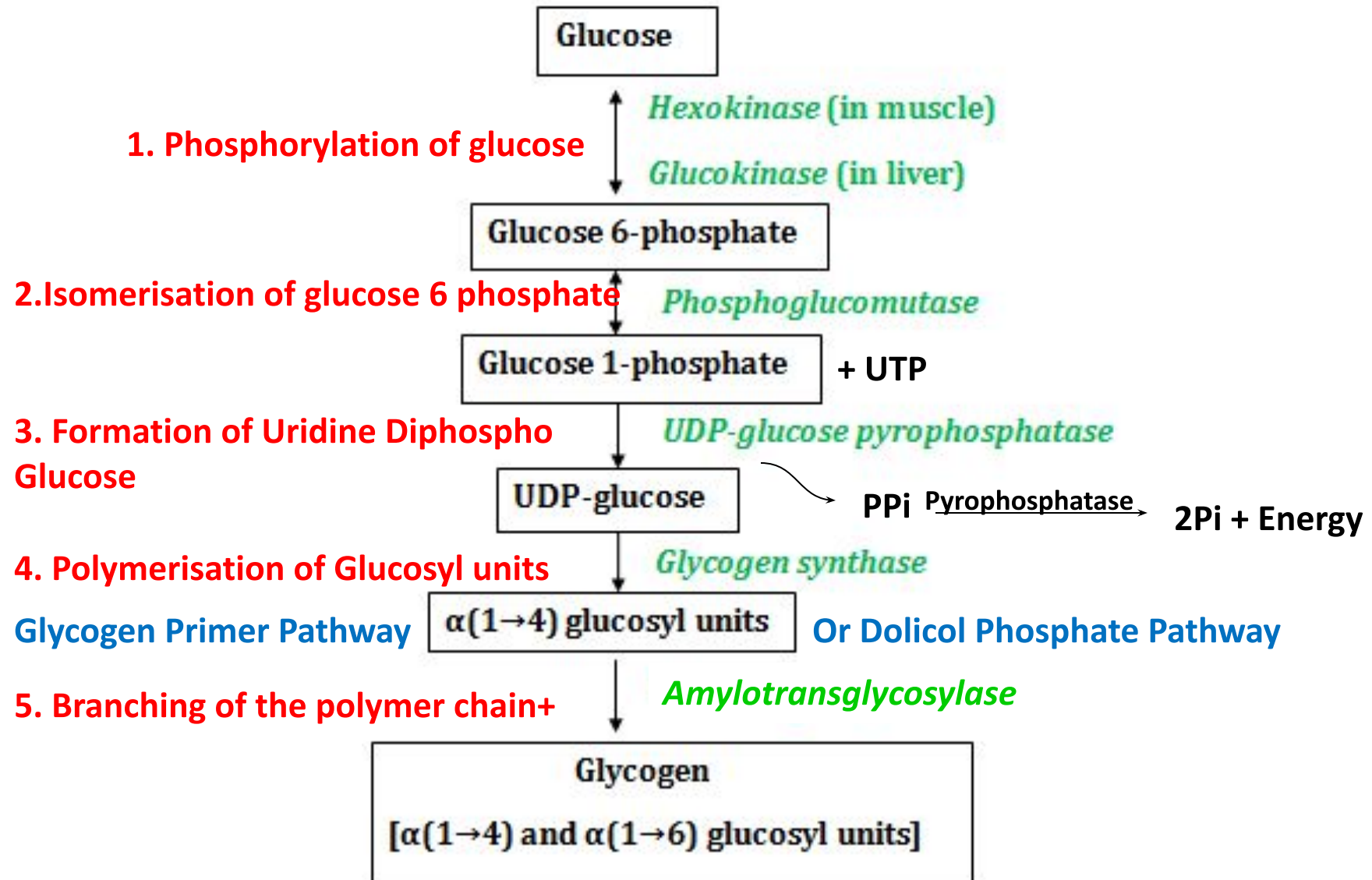
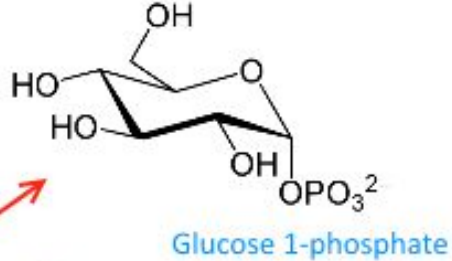
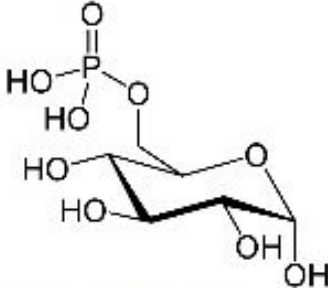
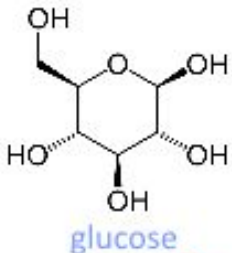
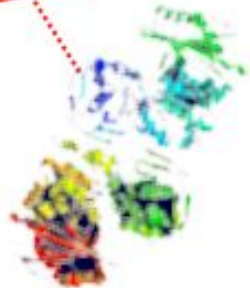


Diagram: Steps of glycogenesis

Glycogenesis



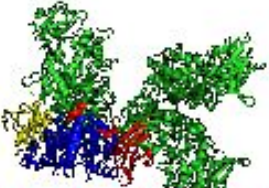
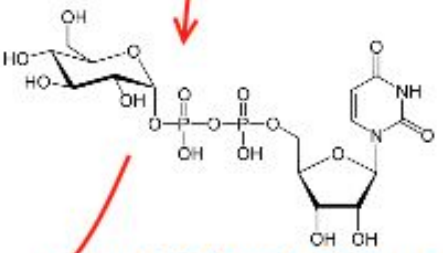
Glucokinase (hexokinase 4)



Phosphoglucomutase



UDP-glucose pyrophosphorylase



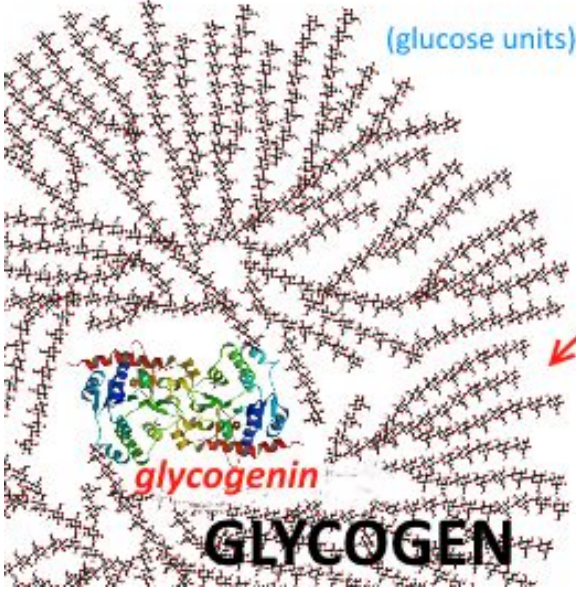
glycogen branching enzyme



glycogen synthase



glycogenin



GLYCOGENOLYSIS

A) PHOPHOROLYSIS

Forms unbranched linear chain

Glucan transferase

Breakdown α 1-6 glycosidic bond

Glucosidase

ATP

Breakdown α 1-4 glycosidic bond

Glycogen Phosphorylase

Pi



Liver

Glycogen

Glucose-1-Phosphate

Phospho glucomutase

Glucose-6-Phosphatase

Glucose-6-Phosphate

Glucose

Glycogen Synthase

Glucose-6-Phosphate

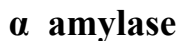
May enter to glycolytic pathway and converted to pyruvic acid

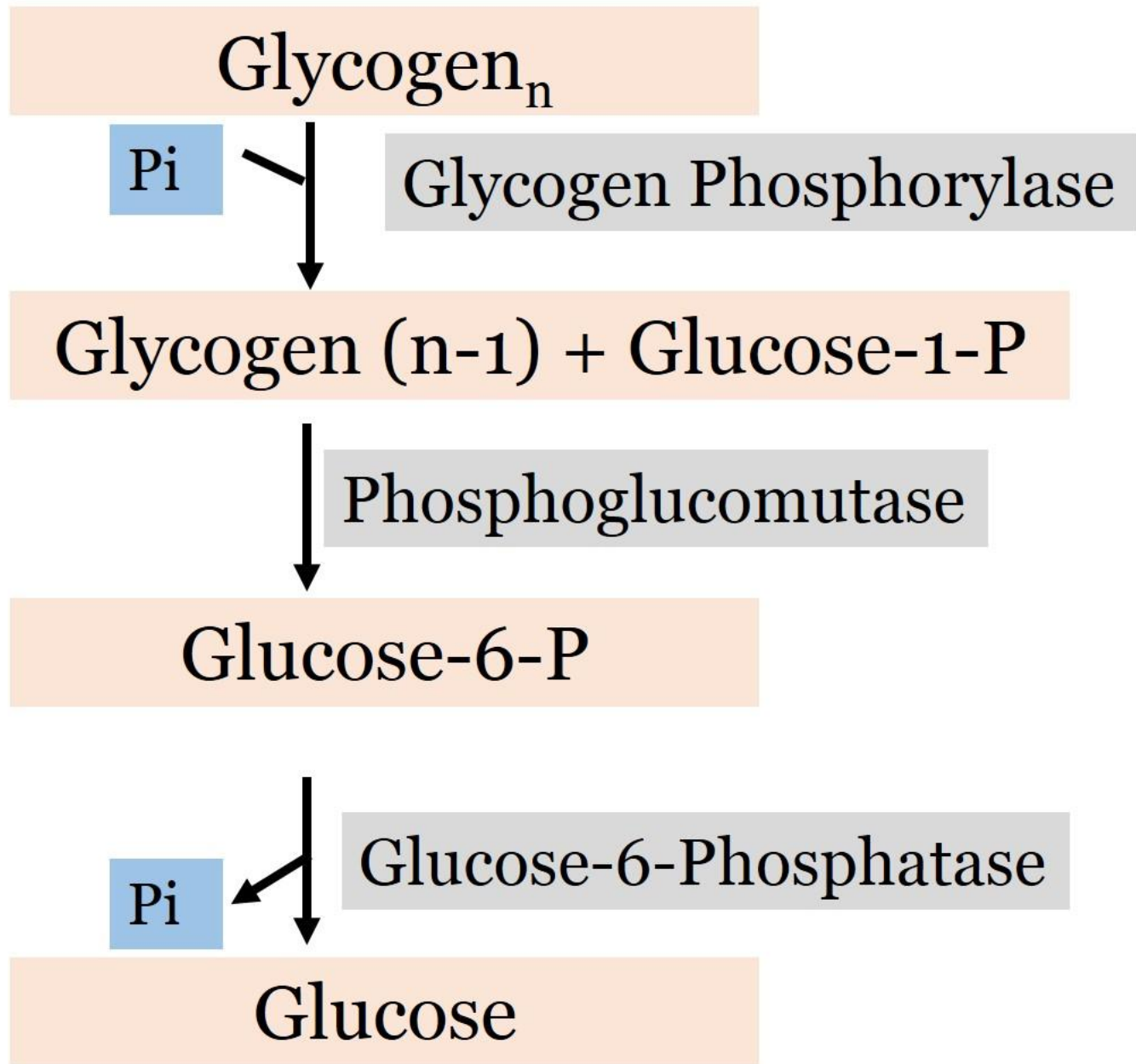
B) HYDROLYSIS / AMYLOSIS

1. Alpha Amylosis

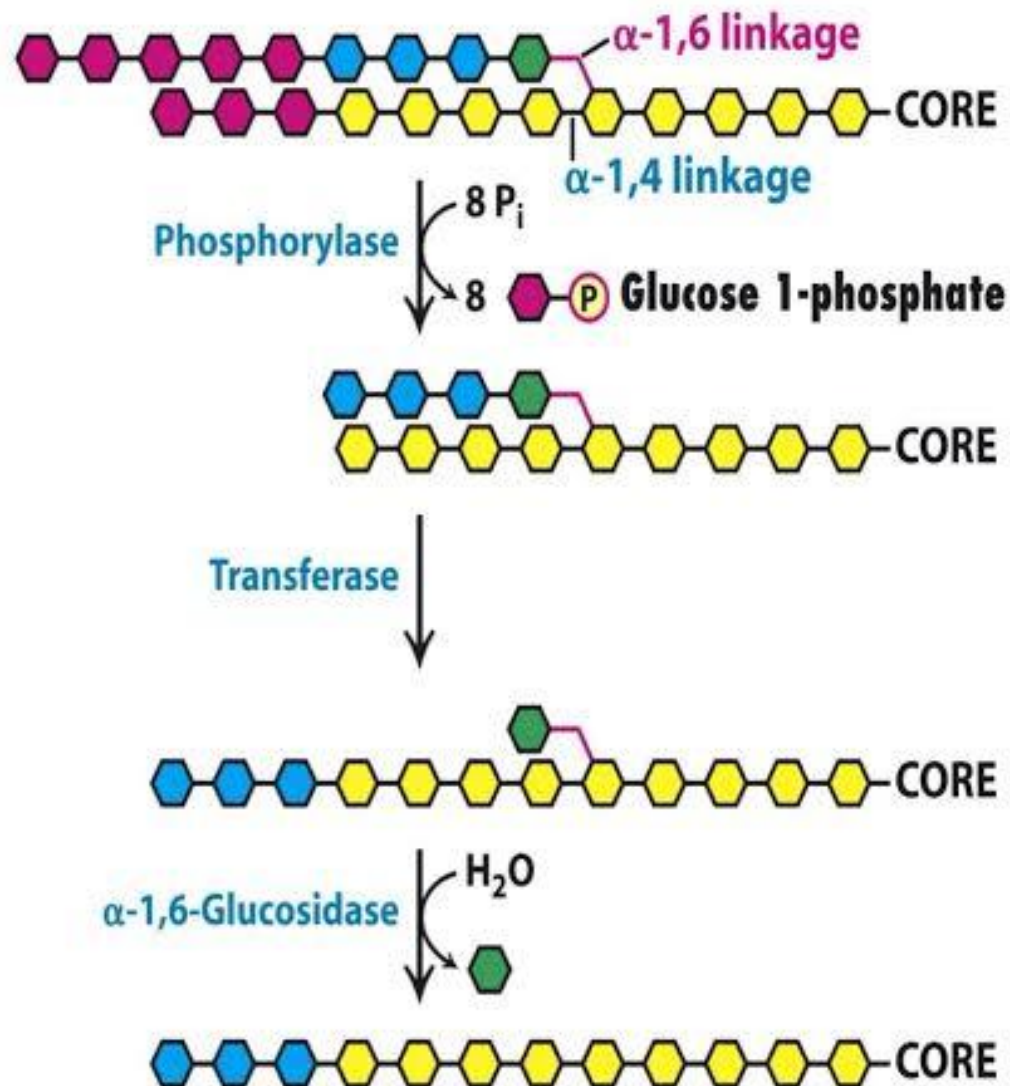


2. Gamma Amylosis



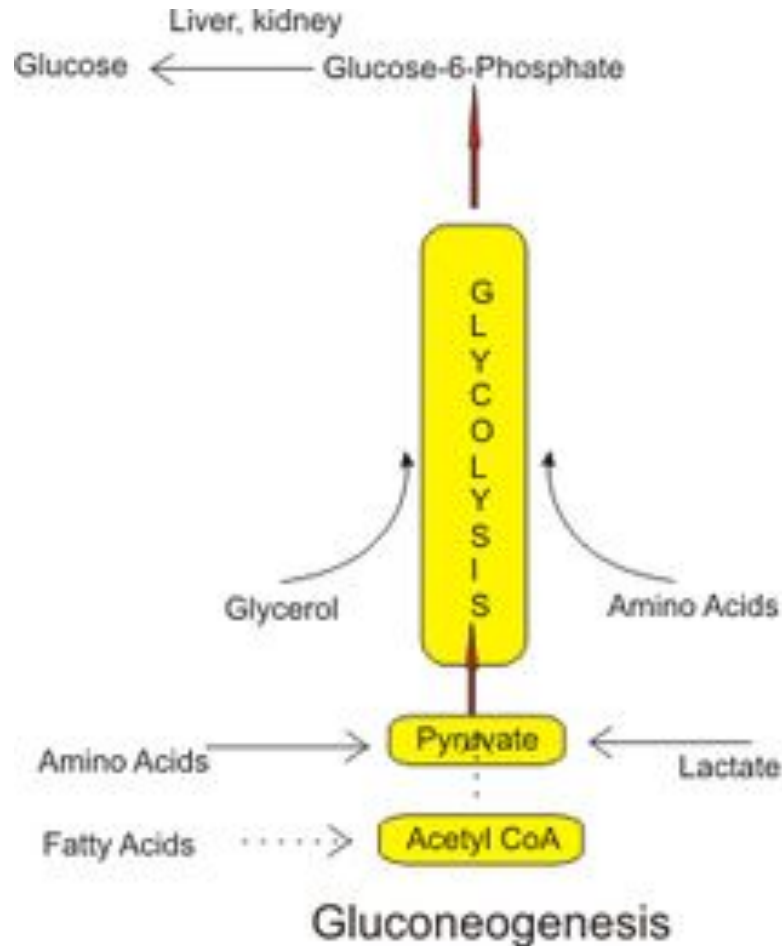


Glycogen degradation cycle



GLUCONEOGENESIS

- Is the metabolic process by which organisms produce sugars (namely glucose) for catabolic reactions from non-carbohydrate precursors



Lactic acid

Lactate Dehydrogenase

Pyruvate

α -keto acids

Pyruvate carboxylase *

Citric acid cycle

Malate Shuttle

Oxaloacetate

Phosphoenolpyruvate

Phosphoenol pyruvate *
carboxykinase (PEPC)

Enolase

2-Phosphoglycerate

Phosphoglycerate mutase

3-Phosphoglycerate

Phosphoglycerate kinase

1, 3-bisphosphoglycerate

Glycerol

Glyceraldehyde 3-phosphate dehydrogenase

Glycerol 3 phosphate dehydrogenase

Glyceraldehyde 3-phosphate

Dihydroxyacetone phosphate

Aldolase

Fructose 1, 6-bisphosphate

Fructose 1, 6-bisphosphatase *

Fructose 6-phosphate

Phosphoglucose isomerase

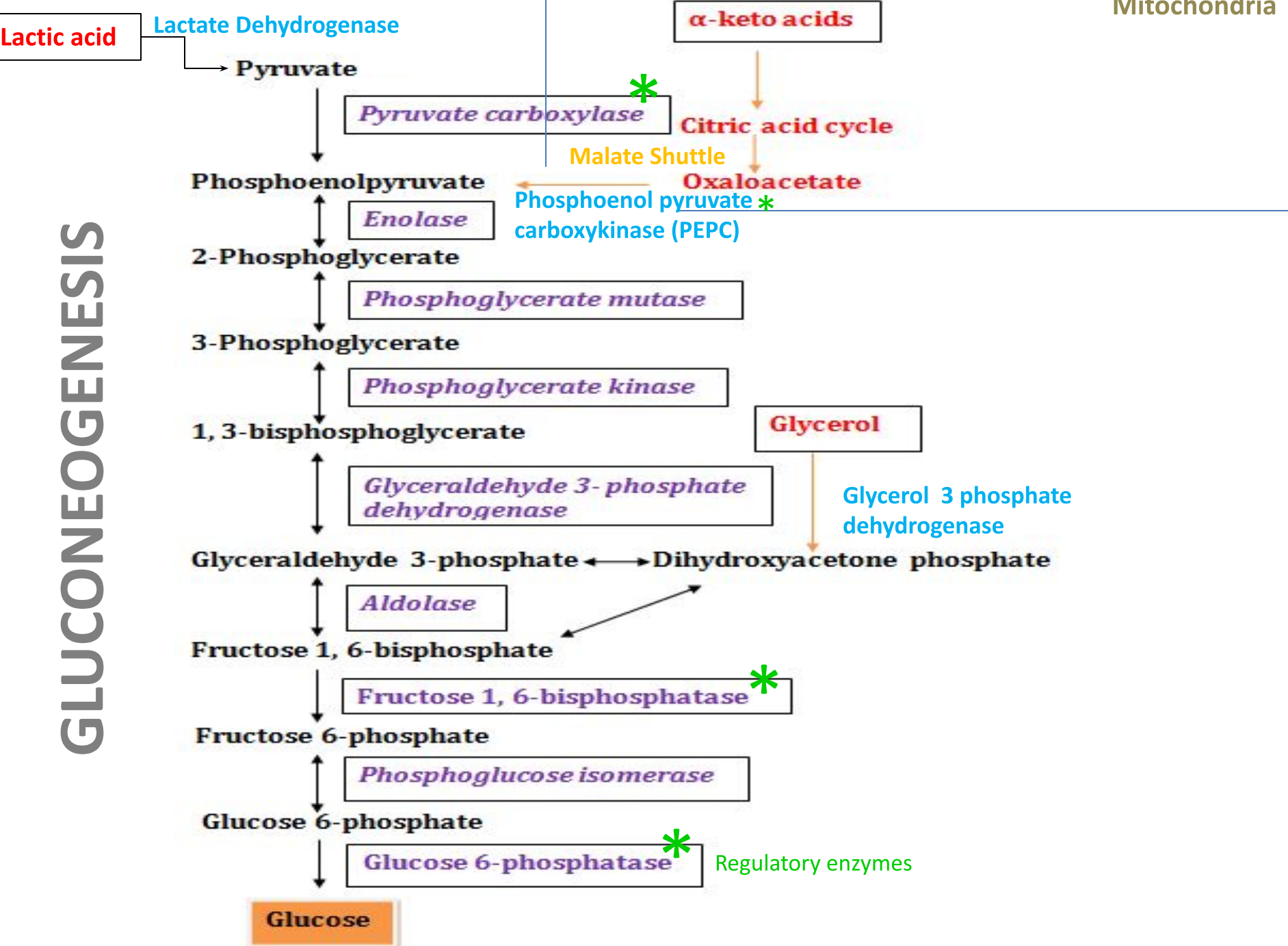
Glucose 6-phosphate

Glucose 6-phosphatase *

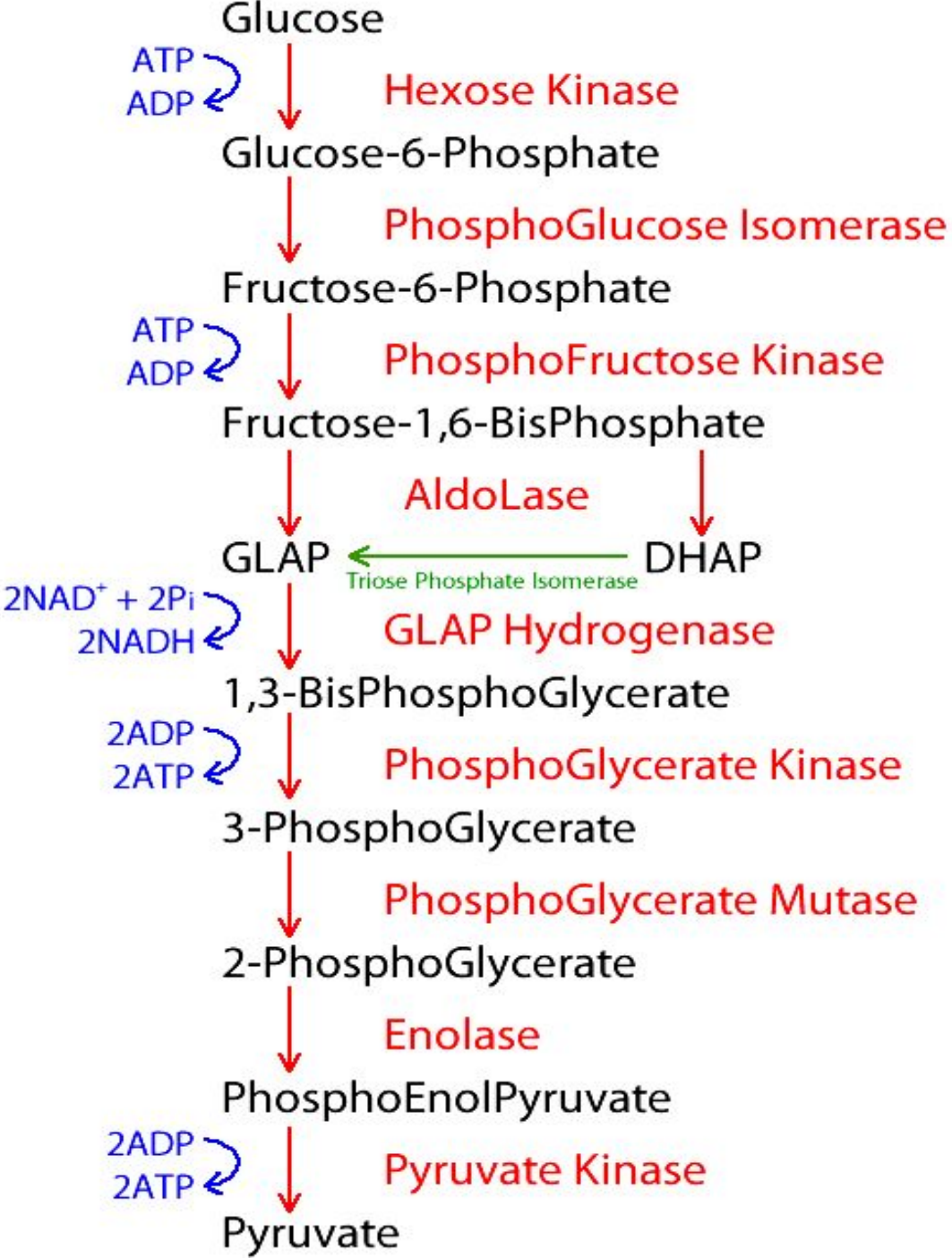
Regulatory enzymes

Glucose

GLUCONEOGENESIS



GLYCOLYSIS



Role of NADH- Bottleneck in glycolysis

- For every glucose molecules in glycolytic pathway 2NAD molecules are reduced to $2\text{NADH} + 2\text{H}^+$ during oxidative phosphorylation of two molecules of glyceraldehyde 3 phosphate.
- The stock of NAD^+ is within the cell is very much limited.
- This problem is solved by the reconversion of NADH to NAD^+ during mitochondrial oxidative phosphorylation.
- In anaerobiosis this is not possible due to absence of oxygen

- In anaerobiosis NAD^+ is uninterruptedly provided by NADH dependent reduction of pyruvic acid to lactic acid in cytoplasm

Conversion of Pyruvic Acid to Lactic Acid

The addition of two H^+ to pyruvic acid forms NADH and lactic acid

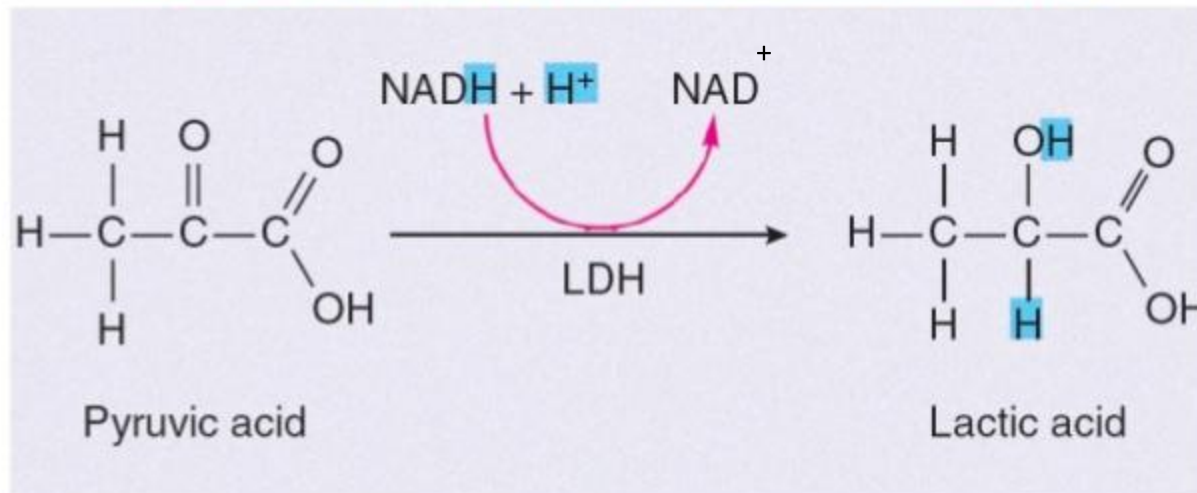


Figure 3.16

Energy balance sheet of Glycolysis

	ATP produced	ATP utilized	Net energy
In absence of oxygen (anaerobic glycolysis)	4 ATP (Substrate level phosphorylation) 2ATP from 1,3 DPG. 2ATP from phosphoenol pyruvate	2ATP From glucose to glucose -6-p. From fructose -6-p to fructose 1,6 p.	2 ATP
In presence of oxygen (aerobic glycolysis)	4 ATP (substrate level phosphorylation) 2ATP from 1,3 BPG. 2ATP from phosphoenol pyruvate.	2ATP -From glucose to glucose -6-p. From fructose -6-p to fructose 1,6 p.	6 ATP Or 8 ATP
	+ 4ATP or 6ATP (from oxidation of 2 NADH + H in mitochondria).		

Importance

- The only metabolic pathway common to all cells
- Only energy yielding process under oxygen free anaerobic condition
- Supplies Carbon skeleton for synthesis of non essential amino acids and glycerol part of alcohol

PASTEUR EFFECT

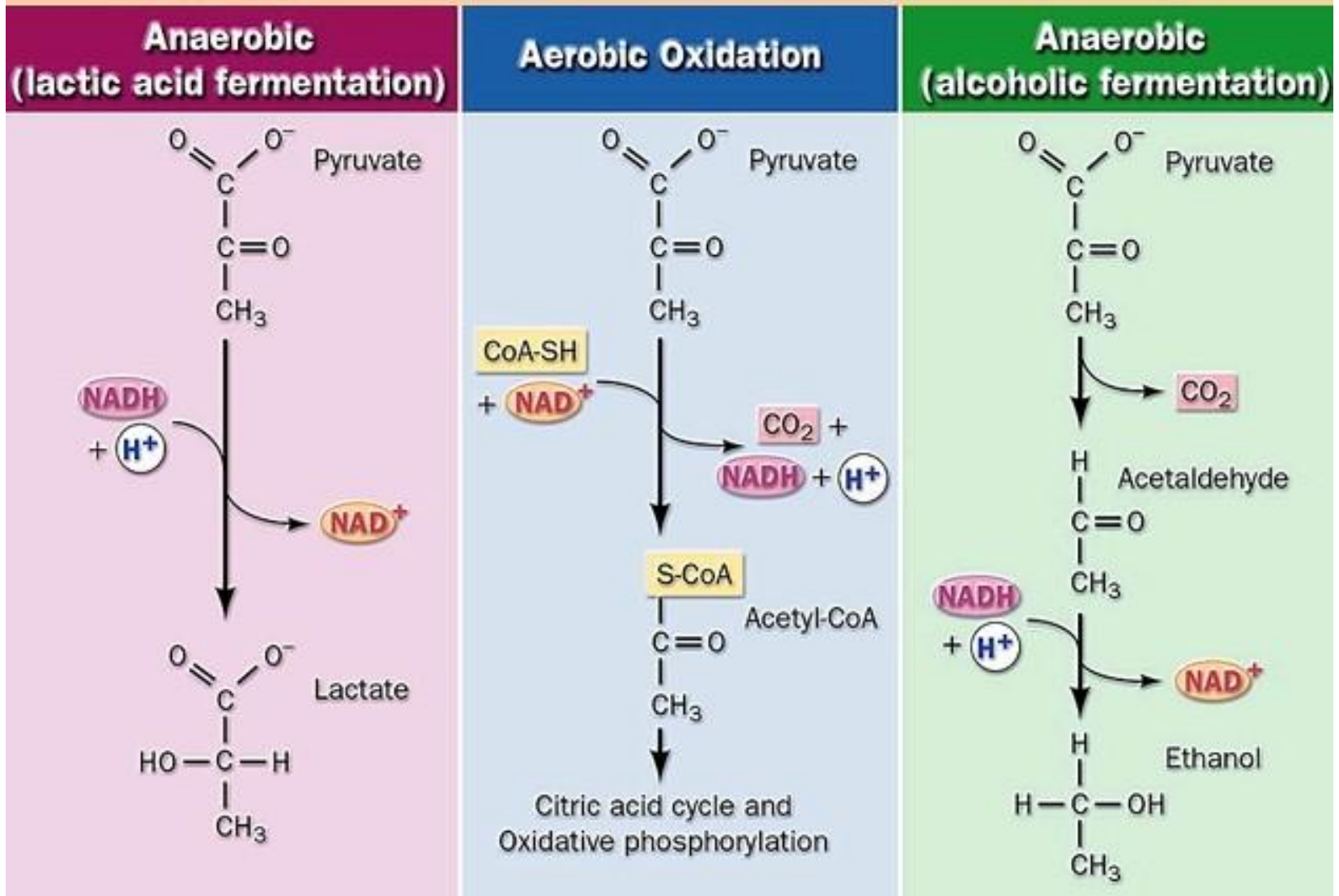
- Suppression of anaerobic respiration by oxygen
- Presence of oxygen increase aerobic respiration – a process that will consume most of the cellular ADP and Pi
- As a result glycolysis and fermentation will not get sufficient supply of them to proceed.
- Further, ADP and Pi are the activators of some glycolytic enzymes and ATP is an allosteric inhibitor of glycolytic enzyme phosphofructo kinase. This all together decrease rate of glycolysis in the cell

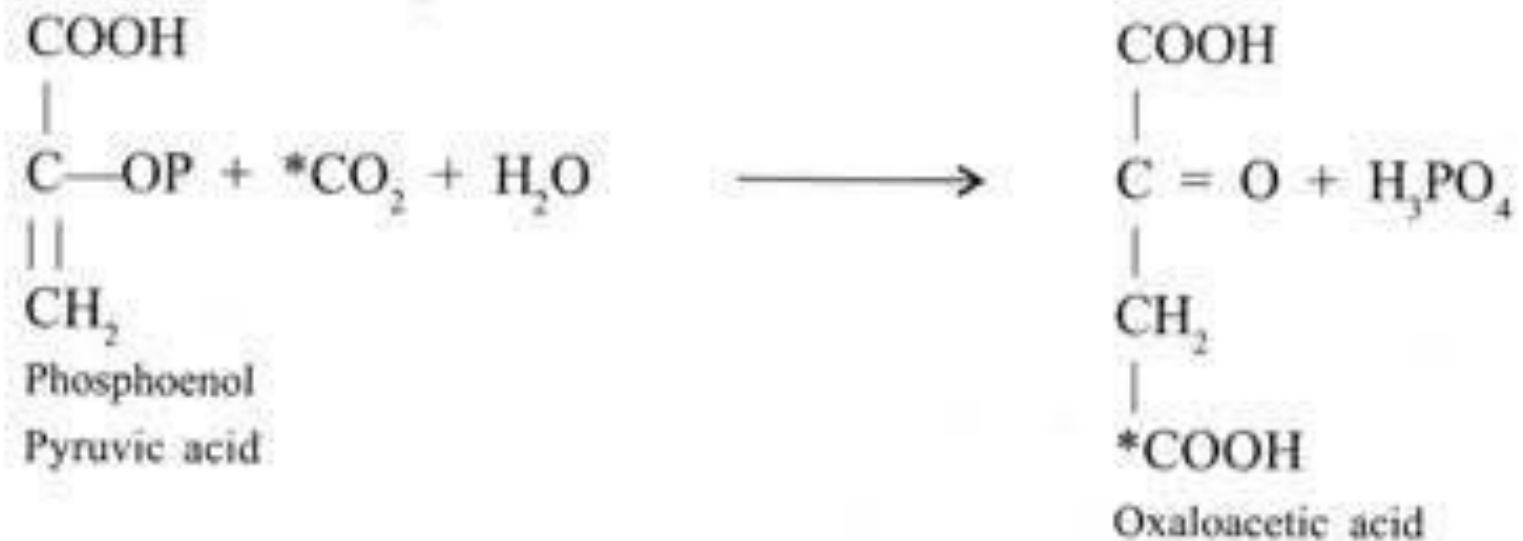
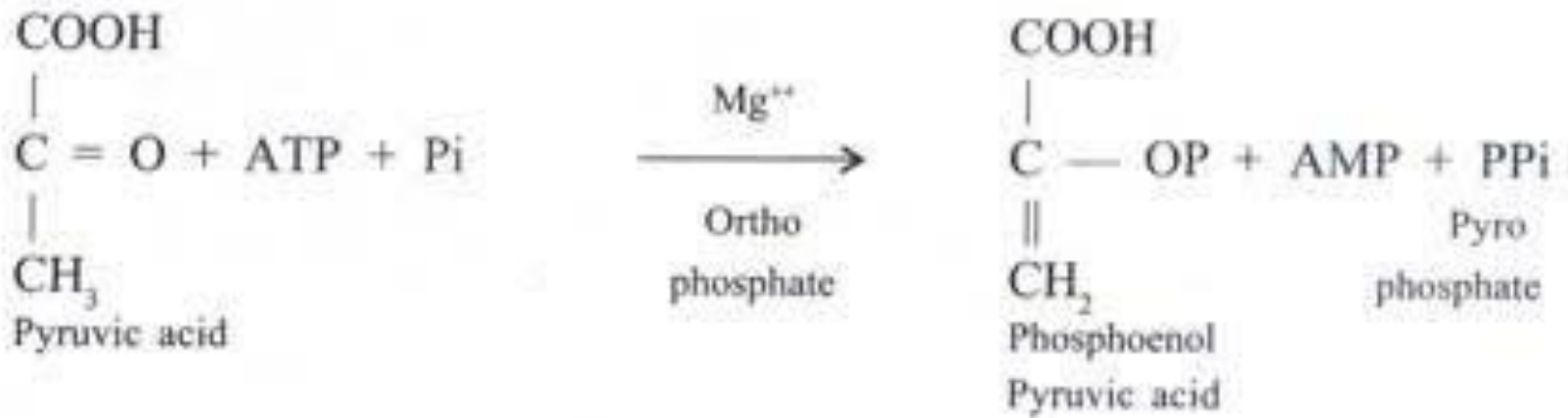
CRABTREE EFFECT

- Reversal of Pasteur effect
- Is the suppression of aerobic respiration by high concentration of glucose
- This is due to increased competition between glycolytic and aerobic process for ADP, Pi and Pi

Fate of pyruvic acid

Three fates of pyruvate produced by glycolysis



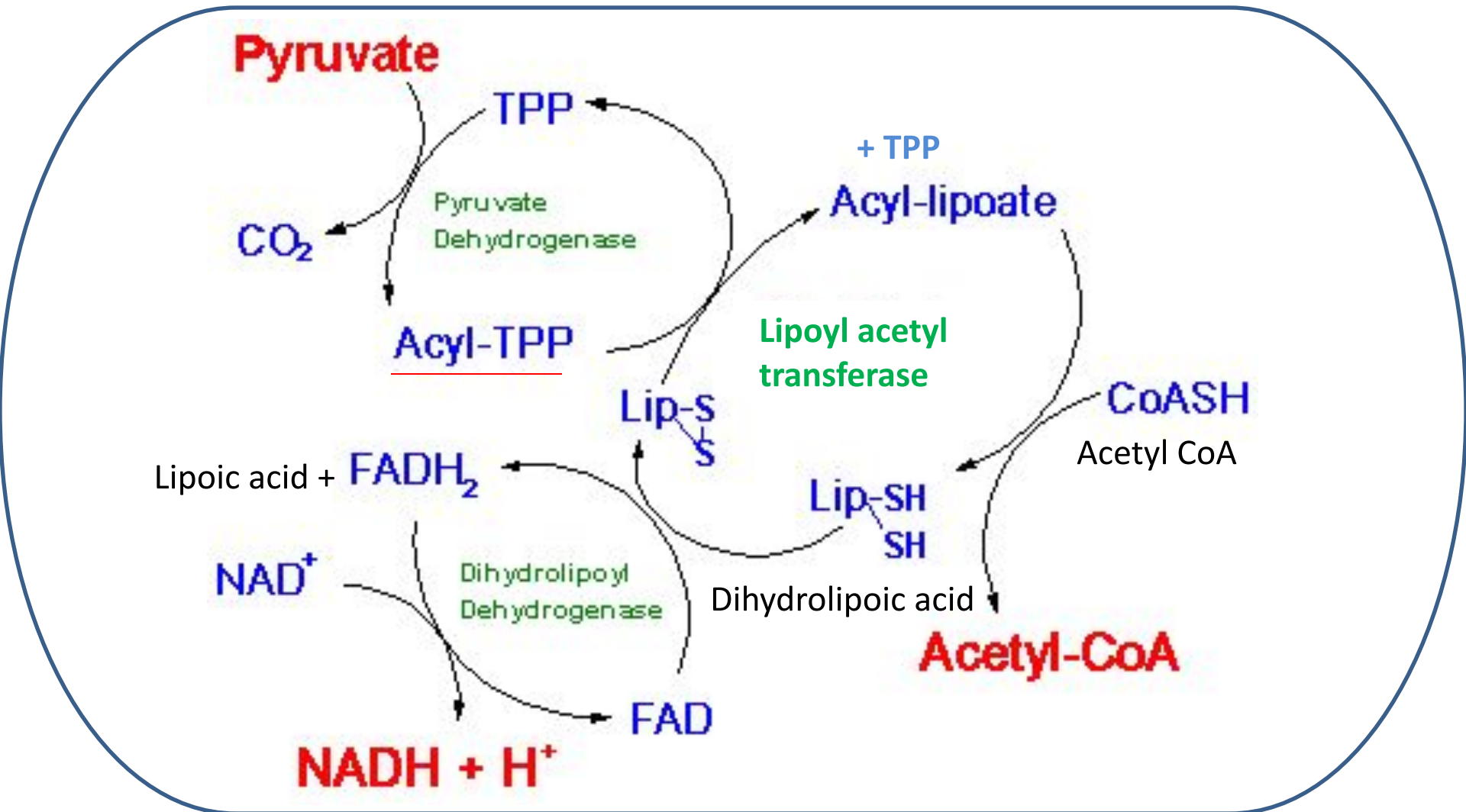


FORMATION OF OXALO ACETIC ACID

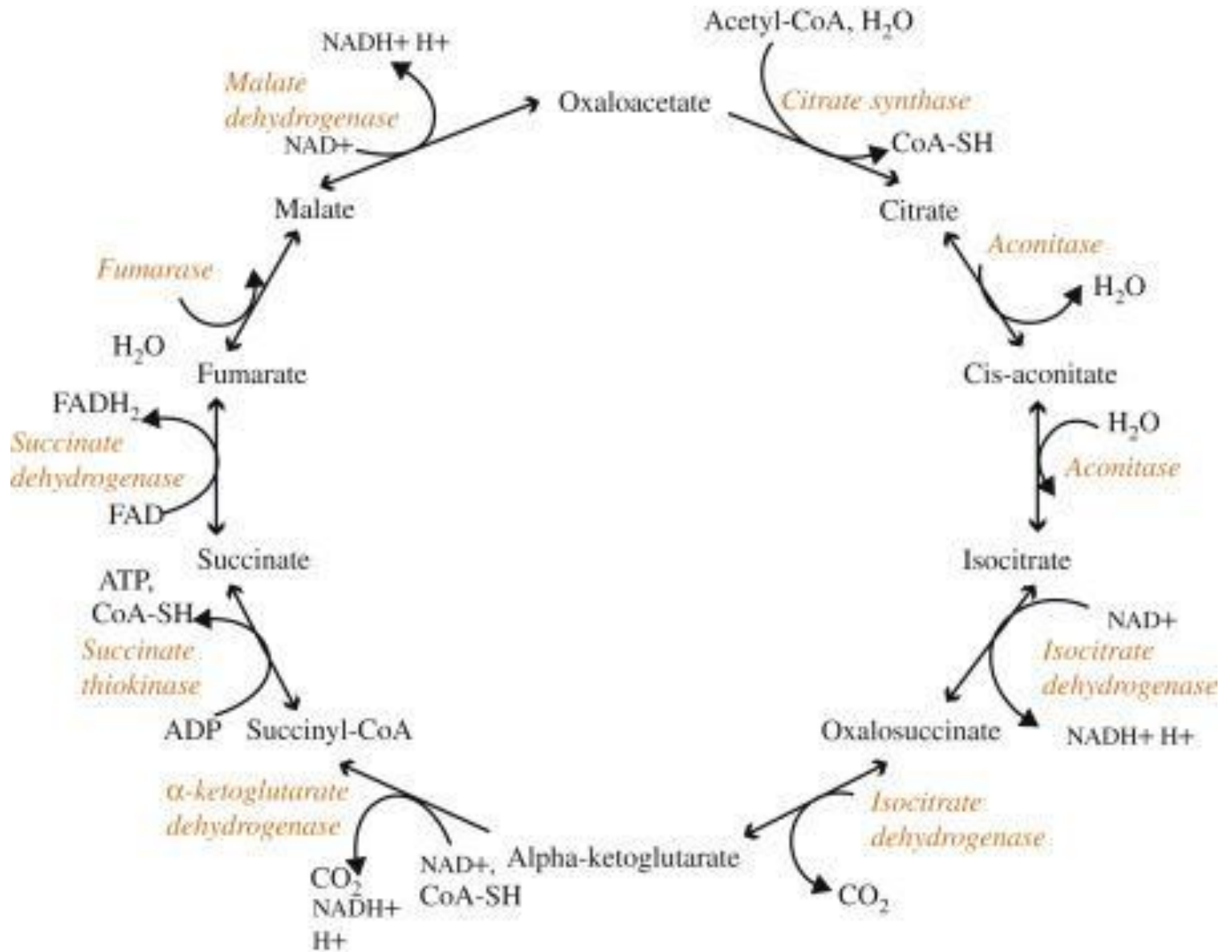
Conversion of Pyruvic acid to Acetyl CoA

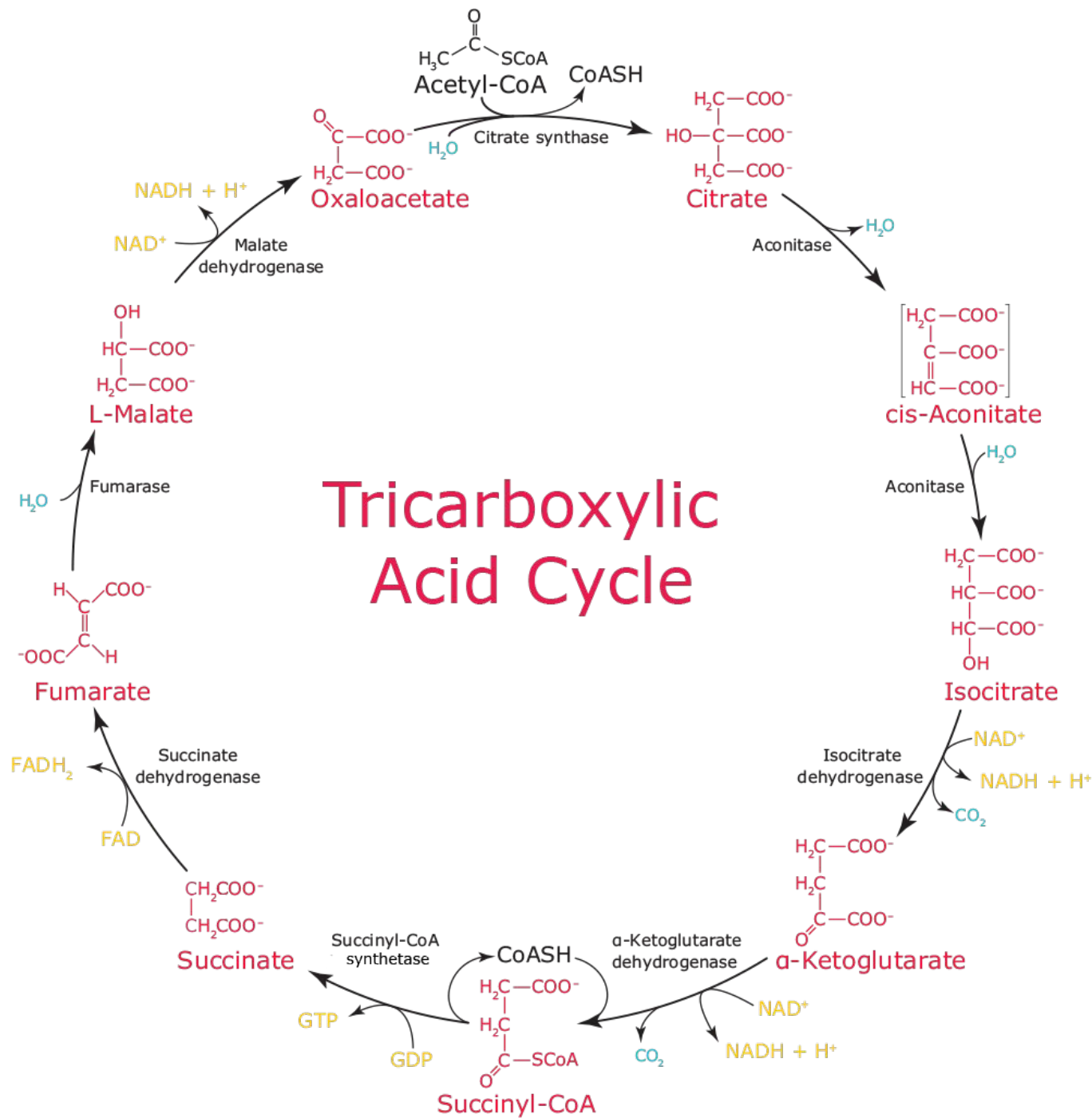
Inside Mitochondria

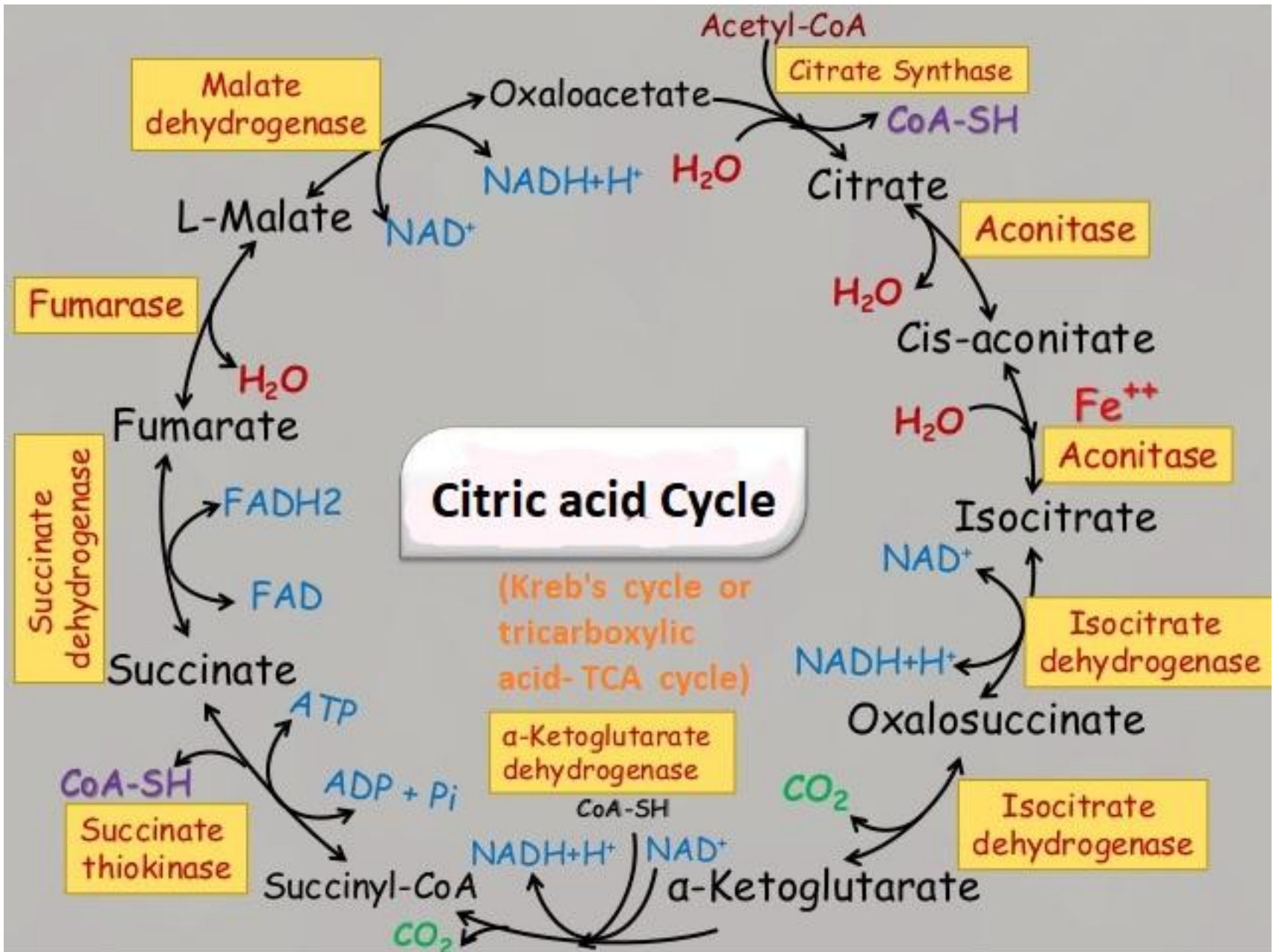
First step of Krebs cycle



KREBS CYCLE









3. Krebs Cycle (Citric Acid Cycle)

- **Total net yield** (2 turns of krebs cycle)
 1. **2 - ATP** (substrate-level phosphorylation)
 2. **6 - NADH**
 3. **2 - FADH₂**
 4. **4 - CO₂**

Energy balance sheet of glucose oxidation

- Old view (theoretical yield)

- **Total ATP Yield**

- 02 ATP** - glycolysis (substrate-level phosphorylation)

- 06 ATP** - converted from 2 NADH - glycolysis

- 06 ATP** - converted from 2 NADH - grooming phase

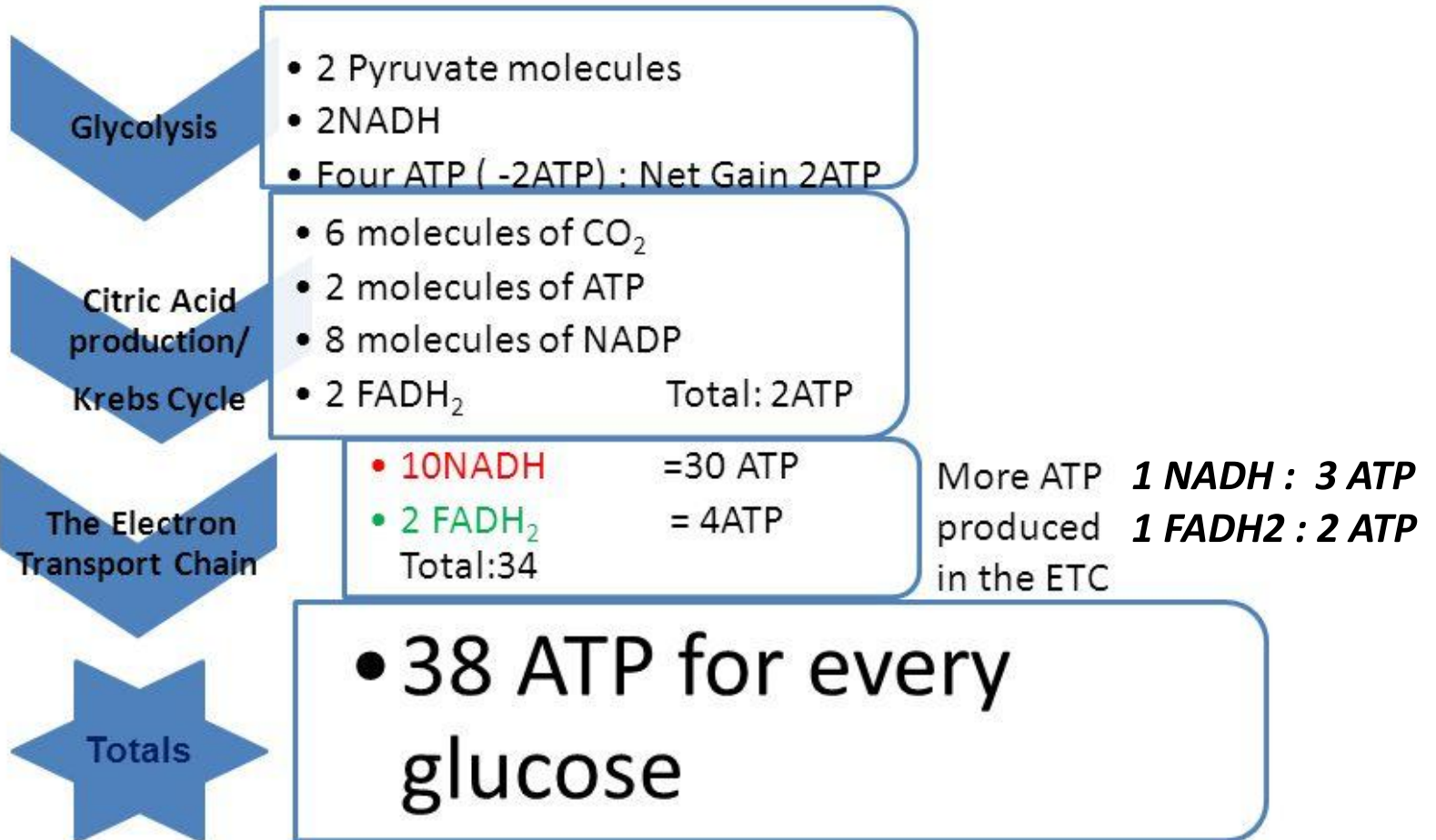
- 02 ATP** - Krebs cycle (substrate-level phosphorylation)

- 18 ATP** - converted from 6 NADH - Krebs cycle

- 04 ATP** - converted from 2 FADH₂ - Krebs cycle

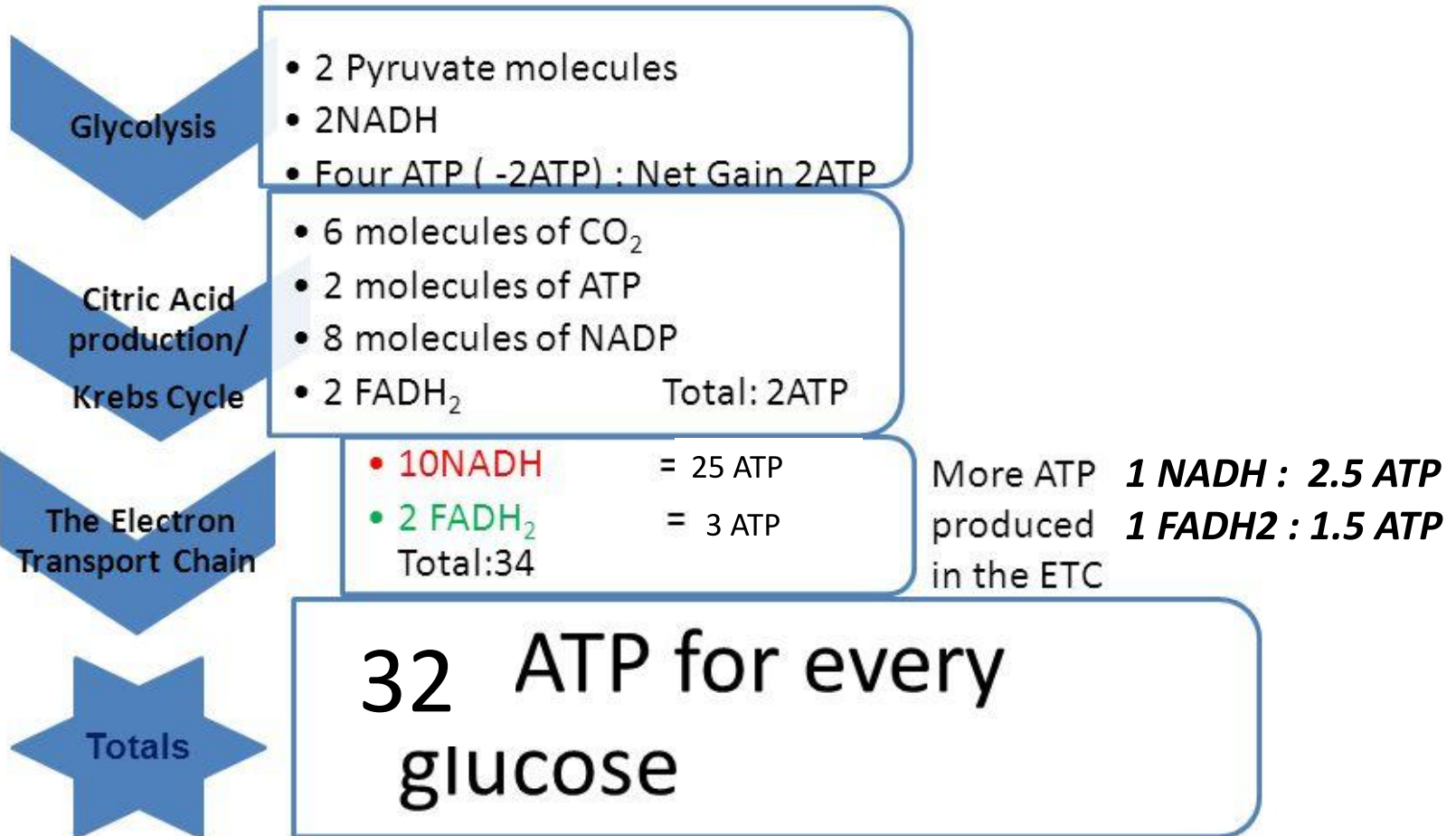
- 38 ATP** - TOTAL

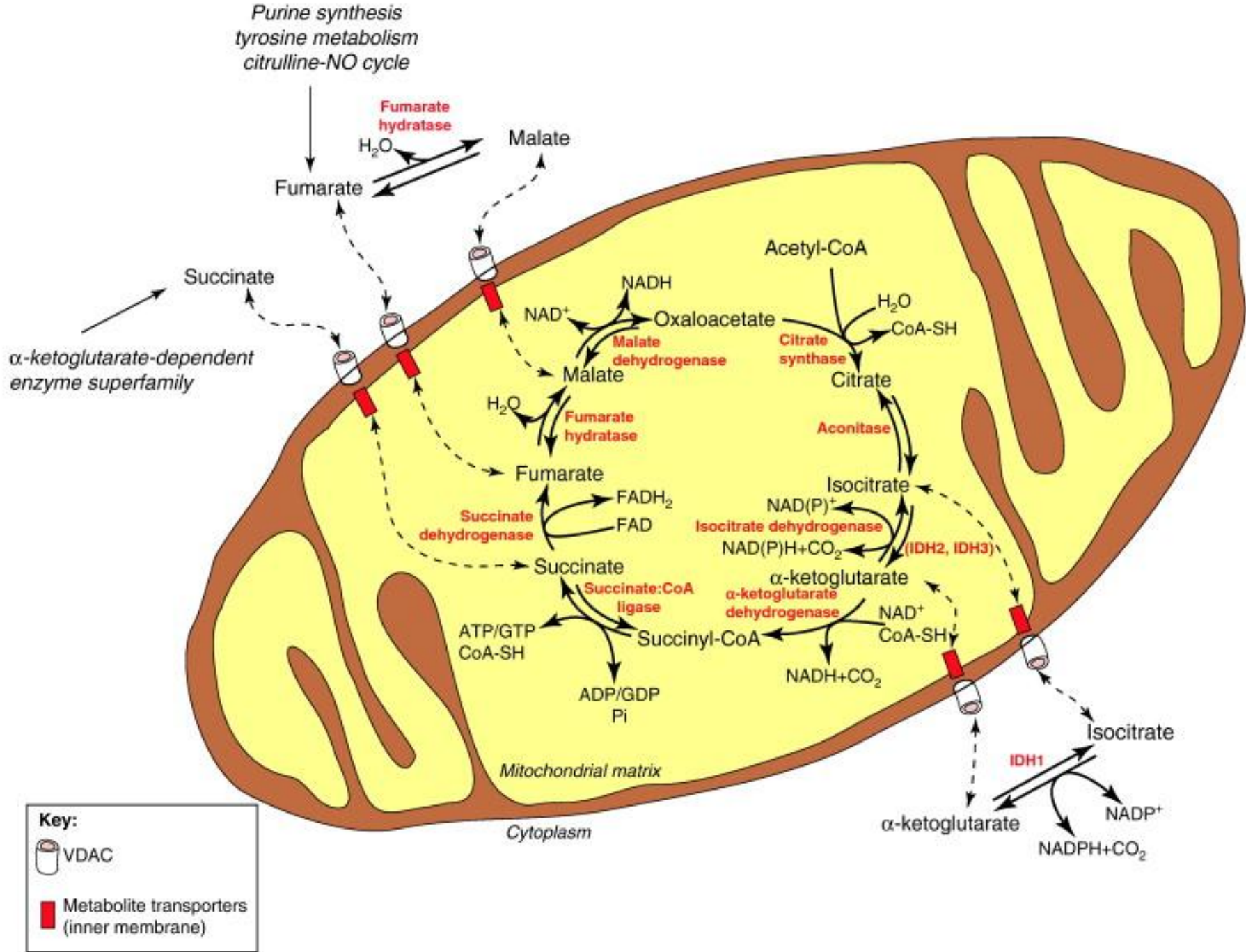
The ATP Totals

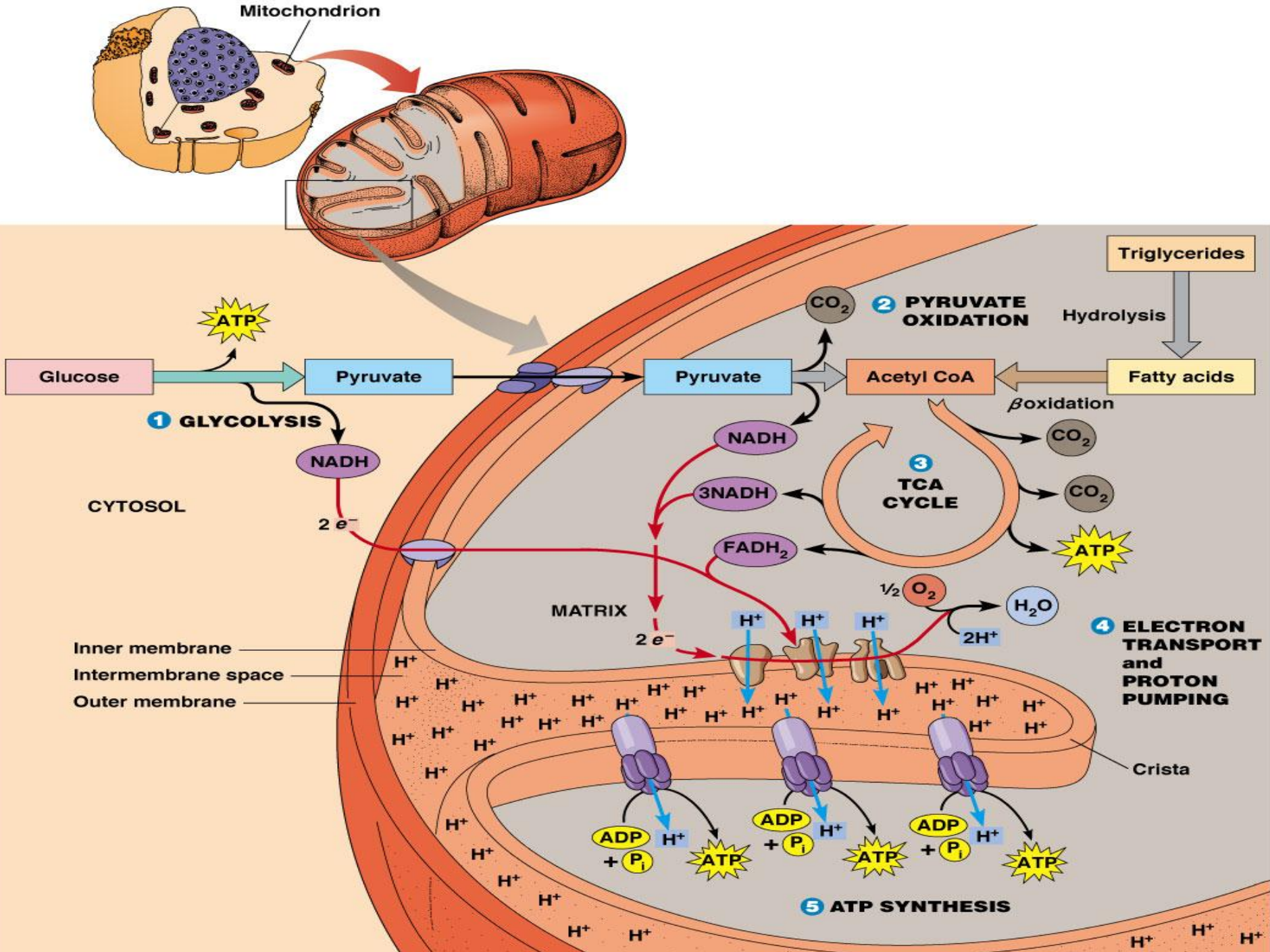


- To pass the electrons from NADH to last Oxygen acceptor, total of 10 protons are transported from matrix to inter mitochondrial membrane
- 4 protons via complex I, 4 via complex III and 2 via complex IV. And to make 1 ATP, 4 protons move from inter mitochondrial membrane to matrix via ATPase
- Thus for NADH— $10/4=2.5$ ATP is produced actually. Similarly for 1 FADH₂, 6 protons are moved so $6/4= 1.5$ ATP is produced.

The ATP Totals

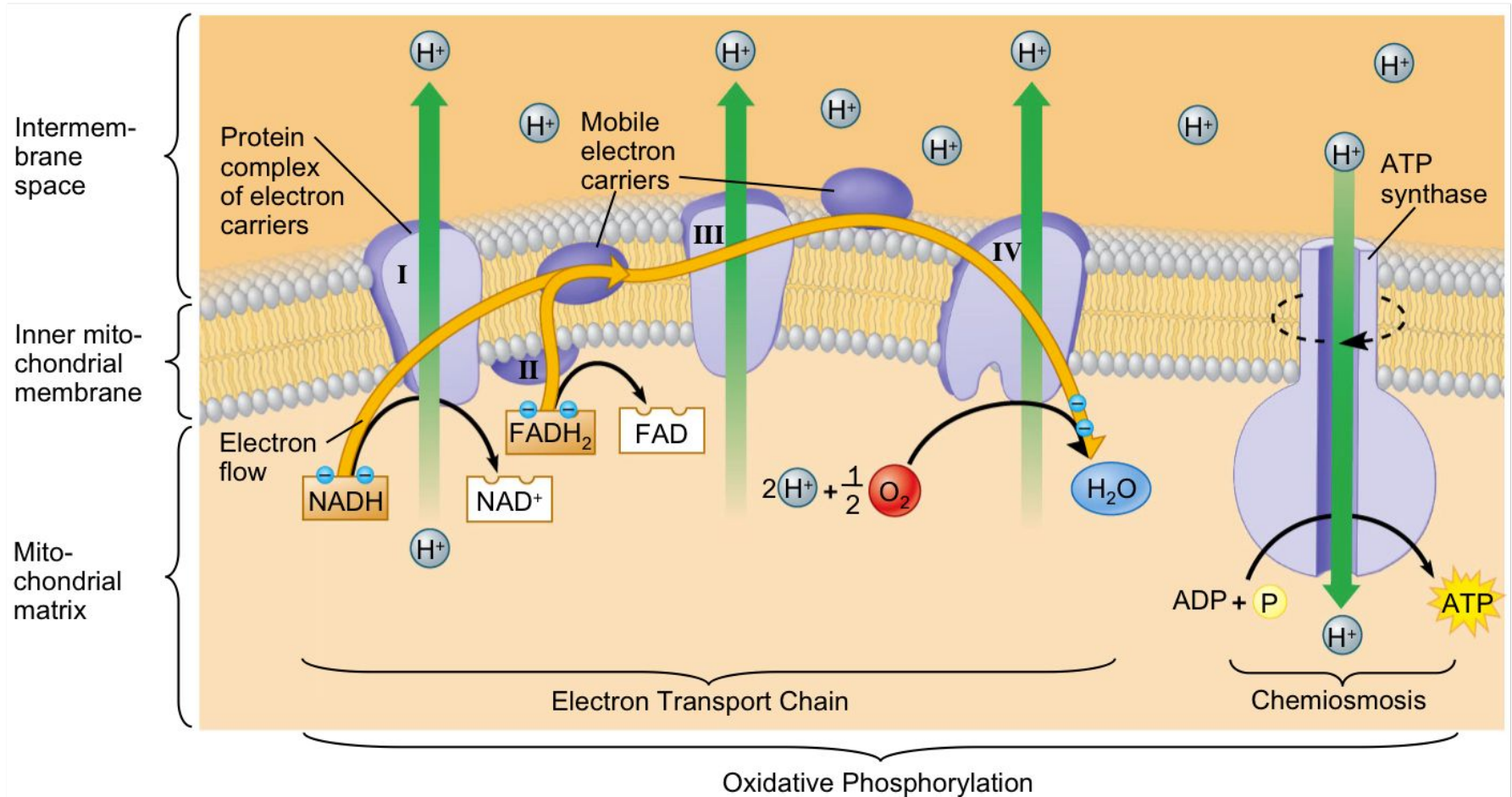


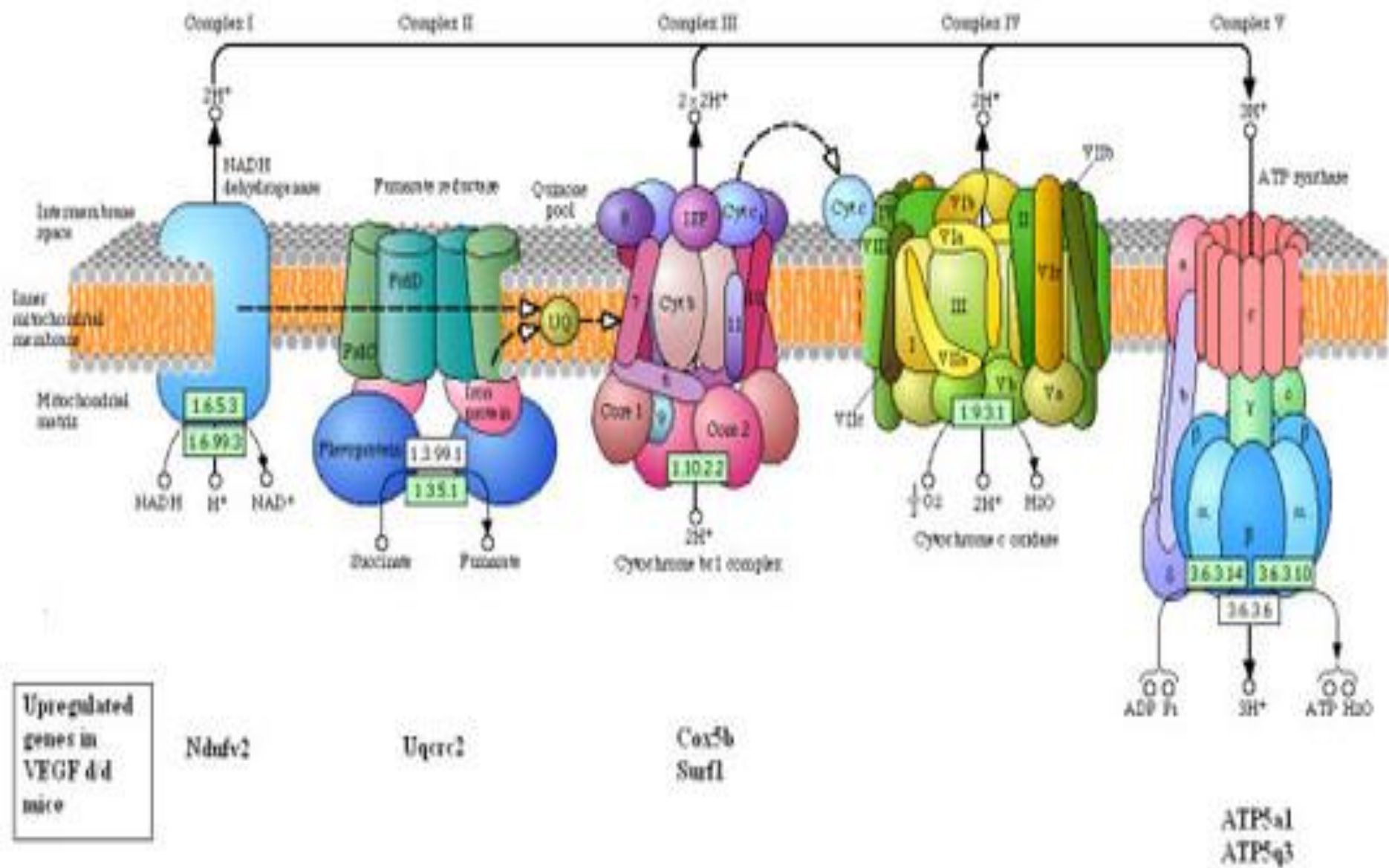




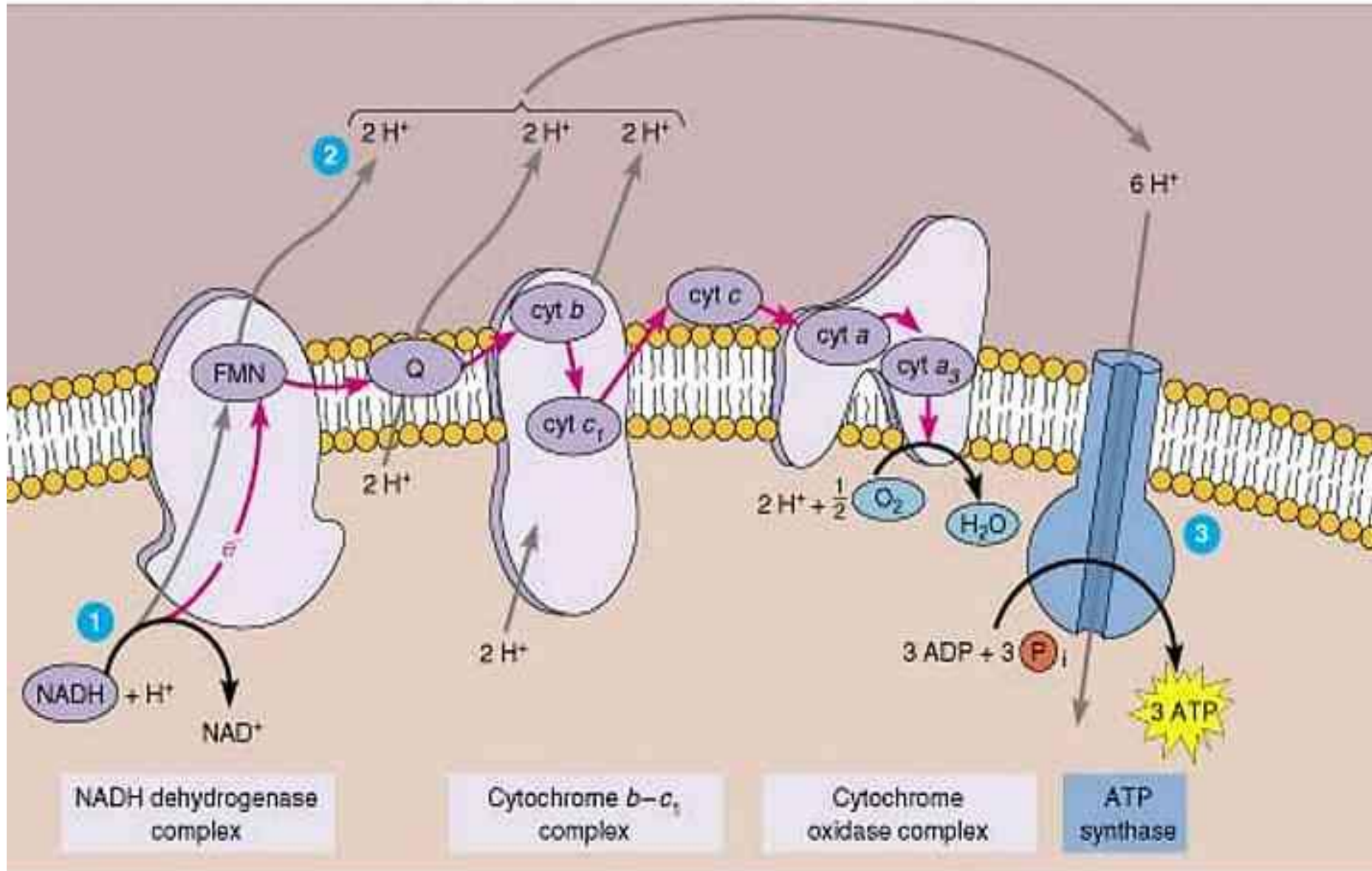
ELECTRON TRANSPORT CHAIN

Figure 6.10



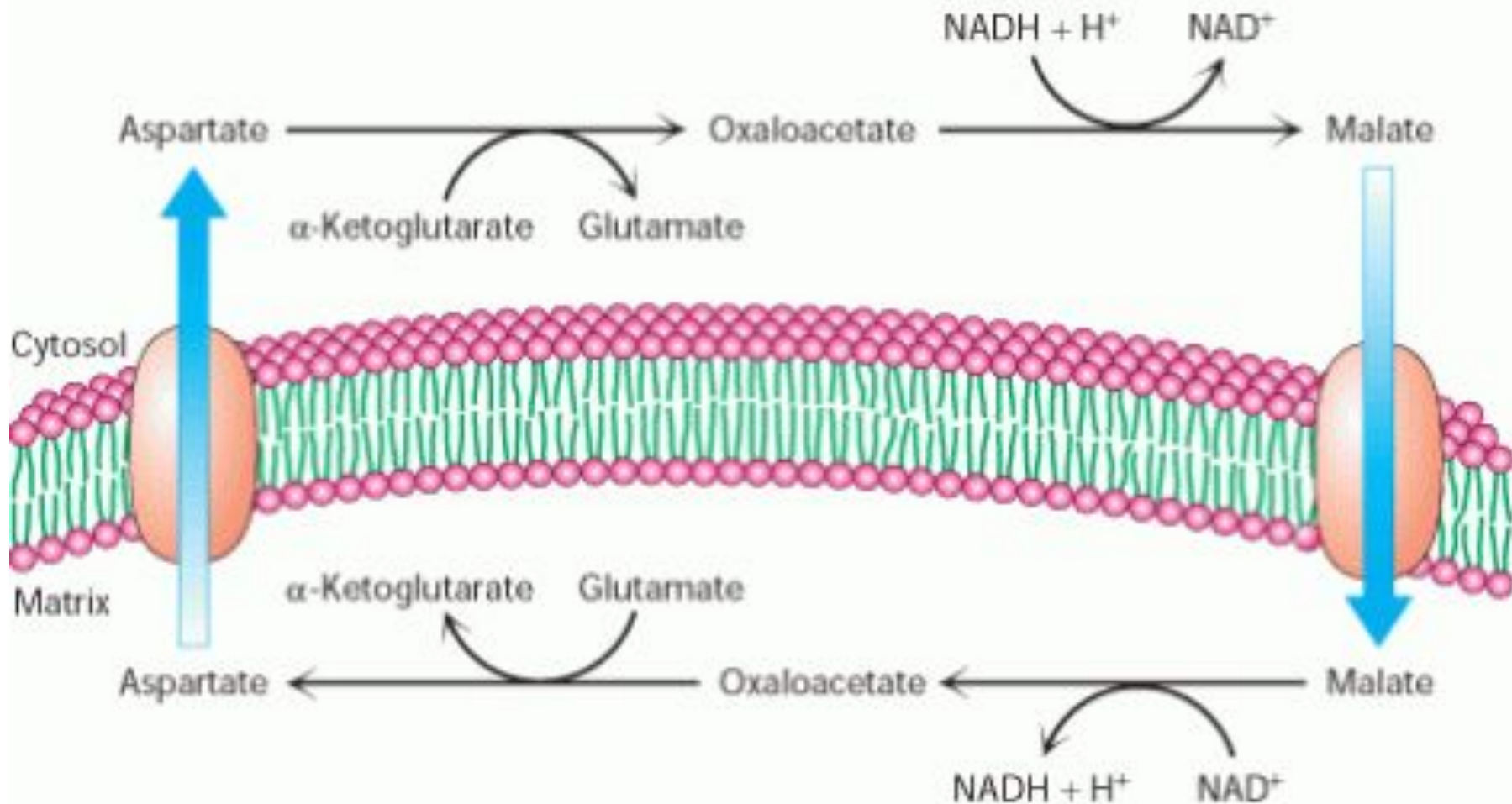


ELECTRON TRANSPORT CHAIN



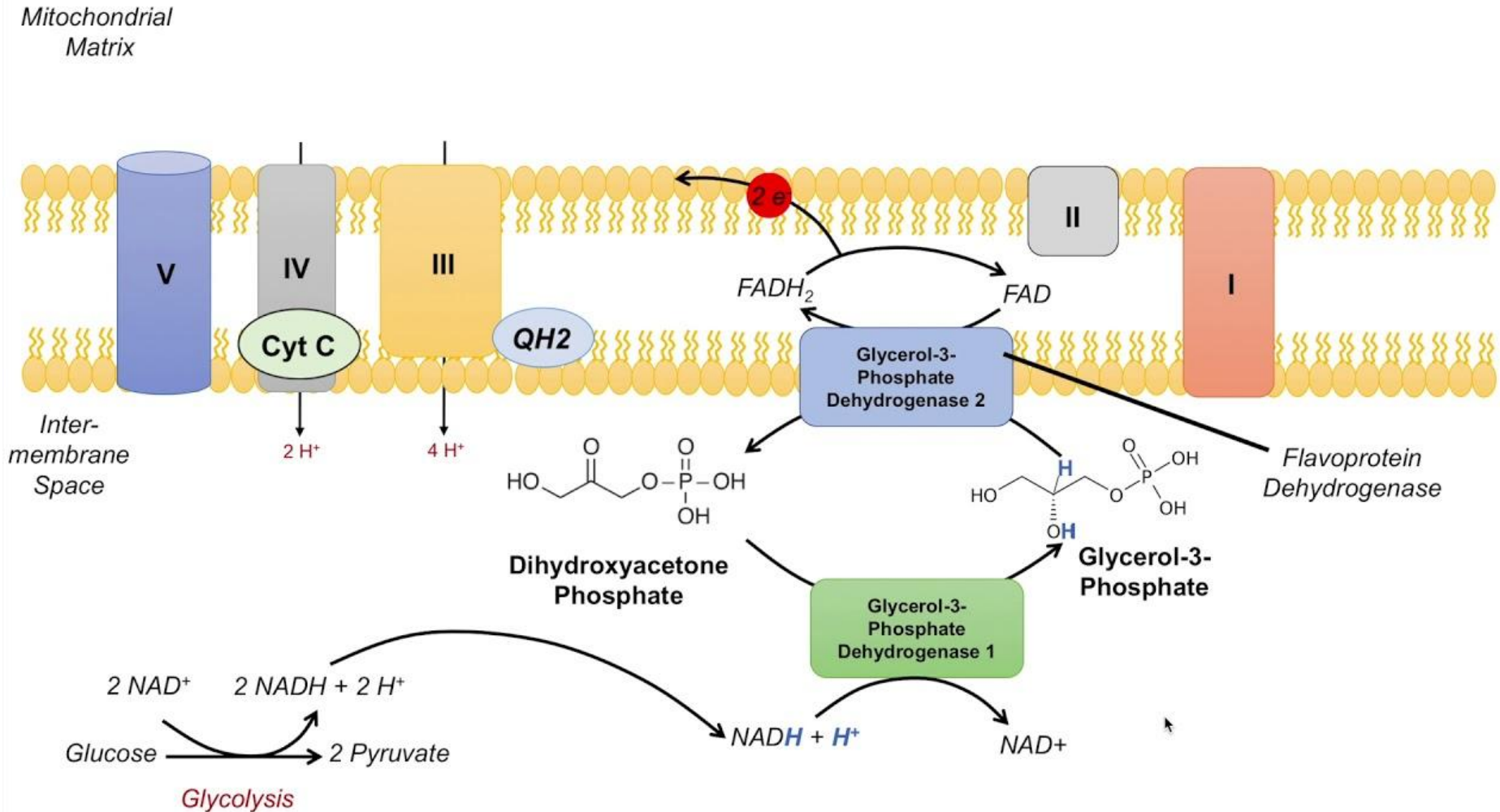
SHUTTLE PATHWAY

(i) Malate- aspartate shuttle



Transport of NADH across inner mitochondrial membrane

(ii) Glycerol 3 phosphate shuttle system



For transport of NADH produced in cytosol by glycolysis